

This tutorial, leading into assignment 2, has two main goals: become proficient at using the Laplace transform; and understanding more about sensors, an important

**Question 1 [4]**

Partial fraction expansion is a topic you have learned about in your prior high school and university math courses.

$$Y(s) = \frac{s + 5}{(s + 1)(s + 4)} = \frac{\alpha_1}{(s + 1)} + \frac{\alpha_2}{(s + 4)}$$

$\alpha_1 =$                        $\alpha_2 =$

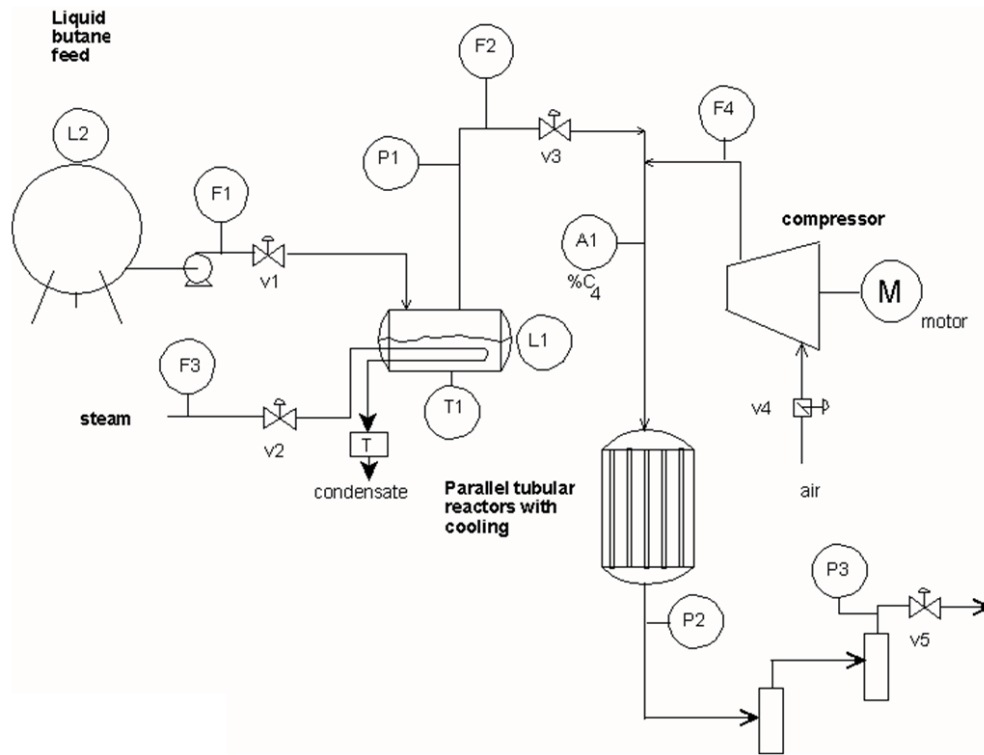
**Question 2 [8]**

Another one:

$$Y(s) = \frac{1}{s(s + 4)(s + 5)} = \frac{\alpha_1}{s} + \frac{\alpha_2}{(s + 4)} + \frac{\alpha_3}{(s + 5)}$$

$\alpha_1 =$                $\alpha_2 =$                $\alpha_3 =$                $\mathcal{L}^{-1}[Y(s)] = y(t) =$

**Question 3 [8]**



This flowsheet shows the process where liquid butane is vaporized and combined with compressed air. You can see the butane holding tanks at 19 seconds into this YouTube video: <http://bit.ly/19D2xrK>

1. Identify 5 types of sensors in the flowsheet.
2. The aim is to control composition of butane, A1, very close to, but below the combustion limit, in the stream entering the reactor. This is a safety-critical control loop, so you must be able to move away from the combustion limit constraint *fast* if you get close to it! Which valve should you use?

Answers:

1. (a)  
(b)  
(c)  
(d)  
(e)
2. I decide to use value valve \_\_\_\_\_. If we open this valve, there us a causal effect on A1 because it will \_\_\_\_\_ (raise/lower) the percentage of butane entering the reactor. Now explain below, giving several reasons, why you selected the valve.