

This tutorial provides some practice for the first midterm.

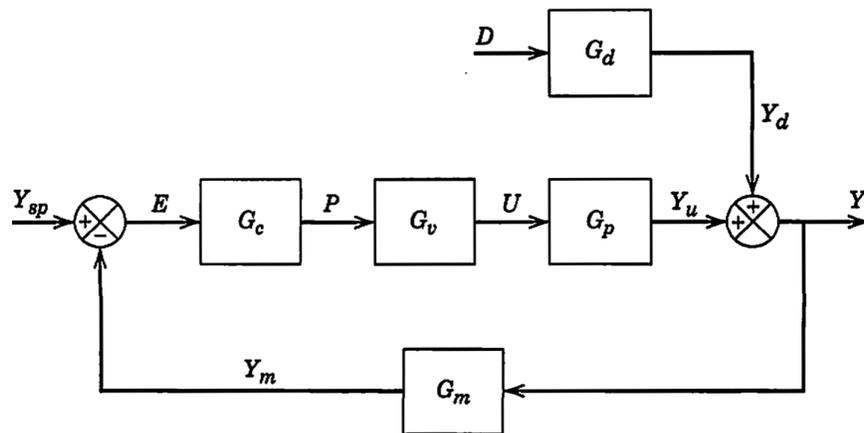
Question 1

Below is the standard feedback control diagram for any process, G_p , which has an input U and output Y_u . There are other transfer functions, such as G_c , which takes the set point value, Y_{sp} and *subtracts* from it the measurement from the sensor, Y_m . The controller then issues a signal to the valve to open or close, via the value P . The valve also has a (fast) transfer function, G_v , which creates the actual process input signal U .

Our measurement on the process, Y , is made up of two parts, Y_d and Y_u . The transfer function G_m represents the dynamics of the sensor (measurement), a similar idea to the transfer function for the thermocouple in question 2 of assignment 3.

Each of the transfer functions for the *systems* are in Laplace form, i.e. G_p should actually be written as $G_p(s)$. Each of the transfer functions for the *signals* are in Laplace form, i.e. Y should actually be written as $Y(s)$.

Today's task is simple: become comfortable with the notation in the block diagram - we are going to see and use this many times, which is why we drop off the (s) parts on the *systems* and *signals*.



1. Derive an expression which gives the output $Y(s)$ for a given valve input signal $P(s)$ and a given disturbance signal $D(s)$. Your answer should be $Y(s) = \dots$
2. Derive a single transfer function which gives the output $Y(s)$ for a given input $Y_{sp}(s)$. For this part of the question, assume input D , and consequently Y_d are both zero (for today). Your answer: $\frac{Y(s)}{Y_{sp}(s)} = \dots$

Question 2

1. The continuous stirred tank chemical reactor shown in the figure is to be analyzed. Assumptions and data for deriving a dynamic model are given below.

- i) the tank is well mixed and isothermal ($T = \text{constant}$)
- ii) the density and the heat capacity ($C_p \approx C_v$) are constant and independent of composition
- iii) flow rates in and out and the volume are constant
- iv) the system is initially at steady state
- v) the reaction $A \rightarrow B$ is irreversible
- vi) the feed stream contains the reactant (C_{A0}) and the inhibitor (C_{I0})
- vii) the rate of reaction of A is $r_A = -k_0 e^{-E/RT} C_A / (1 + k_I C_I)$ where C_I is the concentration of inhibitor. The parameters k_0 , E/R , and k_I are constant.

The inhibitor influences the reaction rate but is neither generated nor consumed in the reaction.

Goal: Determine the dynamic response of the concentration of reactant A after the step in the inhibitor feed concentration.

A. (20 points)

1. By applying basic balances, derive the differential equation(s) that describe the dynamic behavior of the inhibitor concentration (C_I) in the reactor for any change in C_{I0} .
2. Express the equation derived in Part A.1 as a linear (or linearized, as needed) equation in deviation variables. Identify the steady-state gain and the time constant.

B. (20 points)

1. From the basic balances, derive the differential equation(s) that describe the dynamic behavior of composition of A (C_A) in the reactor for any change in C_I .
2. Express the equation derived in Part B.1 as a linear (or linearized, as needed) equation in deviation variables. Identify the steady-state gain and the time constant.

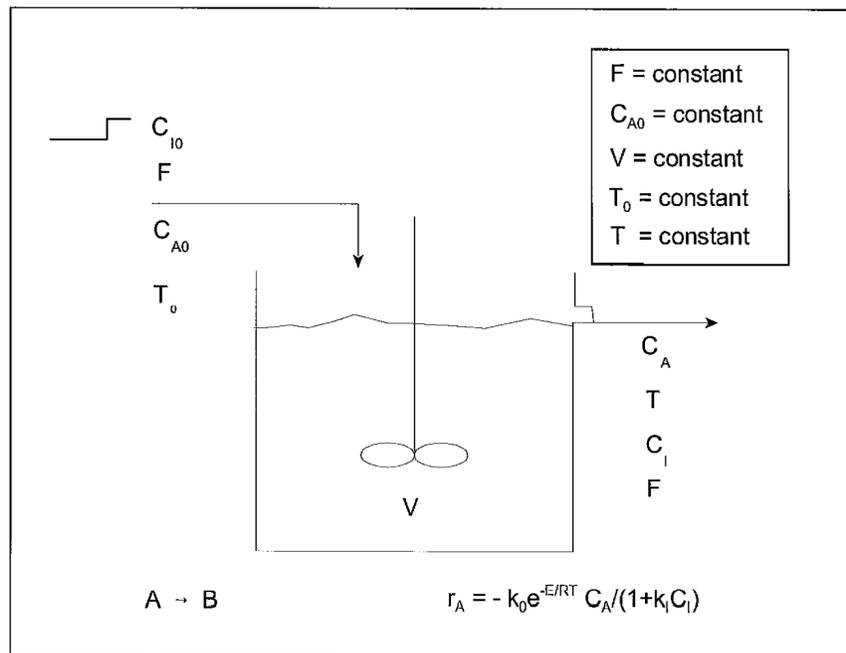


Figure 1. CSTR analyzed in Question 1.