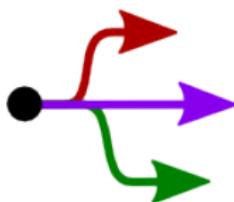


# Separation Processes: Course review

ChE 4M3



© Kevin Dunn, 2014

[kevin.dunn@mcmaster.ca](mailto:kevin.dunn@mcmaster.ca)

<http://learnche.mcmaster.ca/4M3>

Overall revision number: 326 (December 2014)

# Copyright, sharing, and attribution notice

This work is licensed under the Creative Commons Attribution-ShareAlike 4.0 International License. To view a copy of this license, please visit  
<http://creativecommons.org/licenses/by-sa/4.0/>



This license allows you:

- ▶ **to share** - copy and redistribute the material in any way
- ▶ **to adapt** - but you must distribute the new result under the same or similar license to this one
- ▶ **commercialize** - you are allowed to use this work for commercial purposes
- ▶ **attribution** - but you must attribute the work as follows:
  - ▶ “Portions of this work are the copyright of Kevin Dunn”, or
  - ▶ “This work is the copyright of Kevin Dunn”

(when used without modification)

We appreciate:

- ▶ if you let us know about **any errors** in the slides
- ▶ **any suggestions to improve the notes**

All of the above can be done by writing to

kevin.dunn@mcmaster.ca

or anonymous messages can be sent to Kevin Dunn at

<http://learnche.mcmaster.ca/feedback-questions>

If reporting errors/updates, please quote the current revision number: 326

## Course evaluation

Please fill in a course evaluation.

Comments and feedback about the following

- ▶ the assignments
- ▶ course project
- ▶ guest lectures
- ▶ course website, video and audio files
- ▶ extra resources posted on the course website

Every year get better and better from your comments and ideas.

*Thank you!*

## Final exam details

- ▶ On Tuesday, 09 December 2014
- ▶ 16:00 to 19:00
- ▶ IWC-2
- ▶ There will be 90 marks
- ▶ The number of questions does not matter:  
**time management does**
- ▶ Covers all topics learned
- ▶ Expect combination questions: use multiple parts of the course to solve a problem

# More final exam details

**Note:**

- You may bring in any materials to the exam; any textbooks, any papers, *etc.*
- You may use **any calculator** during the exam.
- You may answer the questions in any order on all pages of the answer booklet.
- *Time saving tip: never repeat the question* back in your answer.
- Please use a problem solving strategy on longer questions that is proven to work; *define, explore, plan, do, and importantly, always check.*
- **If anything seems unclear, or information appears to be incomplete, please make a reasonable assumption and continue with the question.**
- Total marks: 90 marks.
- Total time: 3 hours.

## What's in the exam?

Everything that was covered in class time

- ▶ Guest lectures
- ▶ My lectures
- ▶ Videos shown in class time
- ▶ All “interactive tutorial” (question and answer) classes
- ▶ Short quizzes used in class

## What you may bring to the exam

- ▶ Any notes, assignments, midterms, *etc* that you will feel are helpful
- ▶ Any textbooks and printed materials are allowed
- ▶ Any calculator is allowed

Only limitation: no iPads, laptops, tablets, electronic devices, *etc*

# How to do well in the final exam

- ▶ Repeat the midterm without solutions
- ▶ Redo assignment questions you got wrong
- ▶ More importantly: understand **why** you initially got the question wrong
  - ▶ what concept did you misunderstand?
  - ▶ take time to review that concept(s) again
- ▶ Review questions from
  - ▶ Geankolis [3M] textbook: Centrifuges, filtration, membranes
  - ▶ Seader *et al.*: membranes, LLE, adsorption, drying
- ▶ I have posted practice questions to the course website, including all my prior final exams

# What can I do to study well?

## Some tips from the educational research area:

- ▶ Don't just look at a question/topic and say: "Yeah, I could do that". **Prove that you can.**
- ▶ While you are studying:
  - ▶ can you explain the concept to a study partner without looking at the notes?
  - ▶ can you explain **the approach** you would take to solve a problem?

## Poor students do this\*:

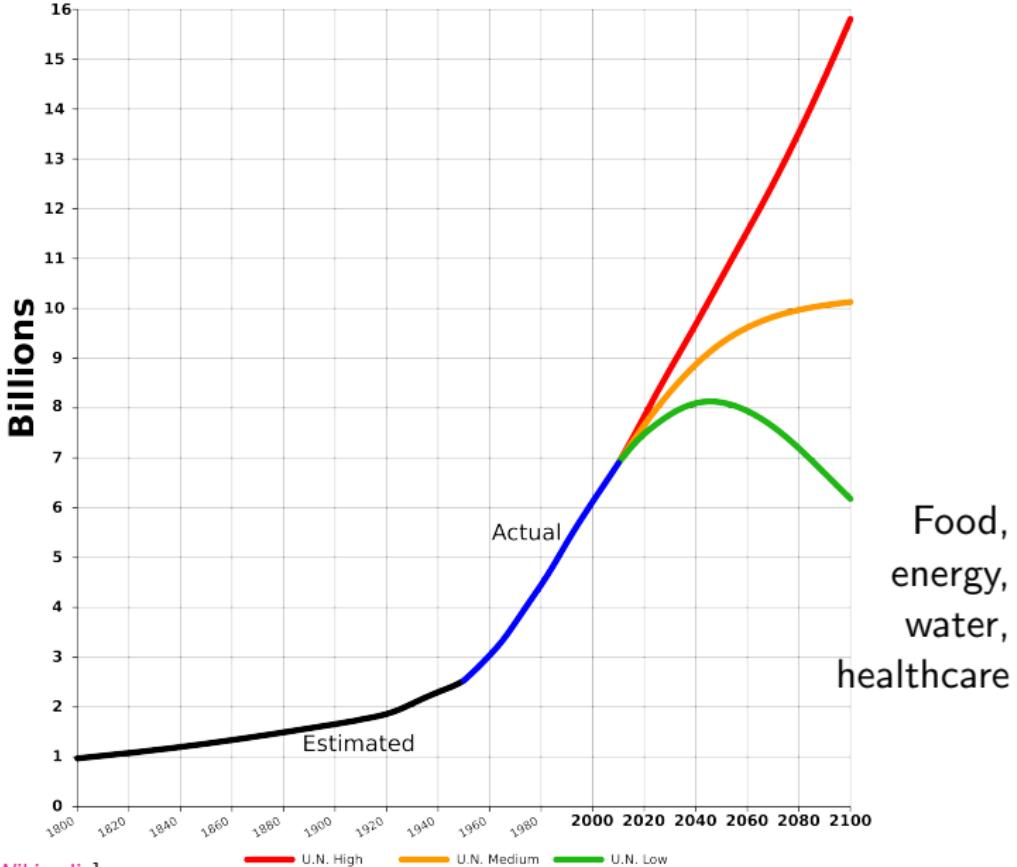
- ▶ Distractions while studying: music, cellphones and email/website checking, TV on in background
- ▶ Skip over parts you don't understand
- ▶ After reading a text, try to repeat it back exactly

\* Meneghetti et al. "Strategic knowledge and consistency in students with good and poor study skills", *European Journal of Cognitive Psychology*, 19, 2007.

## Why study this course: Separation Processes?

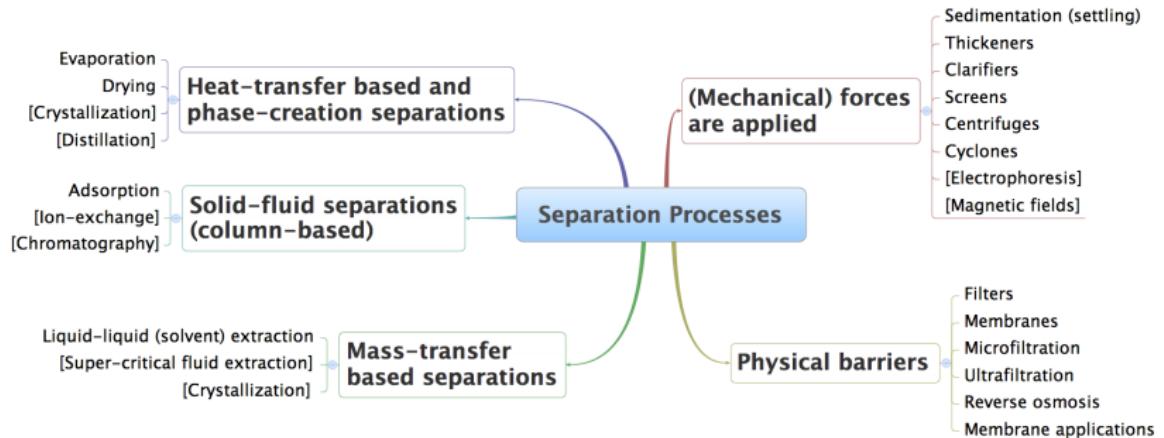
- ▶ Can't beat Nature: "Second Law of Thermodynamics"
- ▶ There are multiple ways to achieve a required separation
- ▶ 50% to 90% of capital investment on petroleum and other chemical-reaction based flowsheets
- ▶ Expense often in proportion to the level of purity (called the separation factor)
- ▶ 60 to 100% of the ongoing operating costs in chemical plants
- ▶ These systems are all around us
  - ▶ leaching (coffee; tea)
  - ▶ centrifugation and drying (washing and tumble drying clothes)
  - ▶ absorption (your lungs)
  - ▶ membranes (your skin, kidneys)
  - ▶ adsorption (water filter)
  - ▶ drying (clothes dryer)

# The course is driven by real needs in society



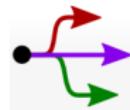
[Loren Cobb, Wikipedia]

# Some more context around the 4M3 course



# Can't remember what was covered when?

Review the subsections on the website



Main page  
Toolbox  
What links here  
Upload file  
Page information

## Adsorption separations - 2014

### References

Read Edit View history Search

- Please use these references to read ahead, or for extra background reading. In alphabetical order:
- Ghosh, R. "Principles of Bioseparations Engineering", Chapter 8, McMaster [\(reserve\)](#)
  - Geankolis, C.J. "Transport Processes and Separation Process Principles", Chapter 12 in 3rd and 4th edition, McMaster Libraries [\(reserve\)](#)
  - Perry's Chemical Engineers' Handbook, Chapter 22, Direct link [\(McMaster subscription\)](#)
  - Richardson and Harker, "Chemical Engineering, Volume 2", 5th edition, Chapter 17 ebook [\(ebook\)](#)
  - Schweitzer, "Handbook of Separation Techniques for Chemical Engineers", Chapter 3.1, McMaster library [\(ebook\)](#)
  - Seader, Henley and Roper, "Separation Process Principles", Chapter 16 in 2nd and 3rd edition McMaster Libraries [\(reserve\)](#)
  - Uhlmanns Encyclopedia article on [Adsorption](#) [\(ebook\)](#)
  - Wankat, "Separation Process Engineering", Chapter 16, McMaster library [\(ebook\)](#)

Date	Class number	Topic	Slides for class	Video and audio files	References and Notes												
12 November	11C	Overview of various adsorbents and their application	<a href="#">Slides</a>	<a href="#">Video</a> <a href="#">Audio</a>													
14 November	11D	Some adsorbents and equilibrium theory	<a href="#">Slides</a>	<a href="#">Video</a> <a href="#">Audio</a>													
19 November	12A	More adsorption case studies, focussing on equilibrium	<a href="#">Slides</a>	<a href="#">Video</a> <a href="#">Audio</a>	Apart from the case studies on adsorption, we consider learning skills, based on the two axes presented in this diagram <a href="#">(ebook)</a> . My challenge to you is to honestly locate your capabilities on this graph, and aim to push yourself to the peak.												
21 November	12B	More details on continuous adsorption with some problems posted in the notes to work through.	<a href="#">Slides</a>	<a href="#">Video</a> <a href="#">Audio</a>	Here are the data from the batch adsorber (slide 32). See if you can calculate the Langmuir isotherm values. <table border="1"><thead><tr><th>C<sub>A</sub></th><th>C<sub>S,A</sub></th></tr></thead><tbody><tr><td>0.004</td><td>0.026</td></tr><tr><td>0.009</td><td>0.053</td></tr><tr><td>0.019</td><td>0.075</td></tr><tr><td>0.027</td><td>0.082</td></tr><tr><td>0.094</td><td>0.123</td></tr></tbody></table>	C <sub>A</sub>	C <sub>S,A</sub>	0.004	0.026	0.009	0.053	0.019	0.075	0.027	0.082	0.094	0.123
C <sub>A</sub>	C <sub>S,A</sub>																
0.004	0.026																
0.009	0.053																
0.019	0.075																
0.027	0.082																
0.094	0.123																

Class date(s): 12 November 2014

### 2014-4M3 Class 11C (12 Nov)

26:24

Download video: [Link](#) [426 M]

### 2014-4M3 Class 11D (14 Nov)

49:42

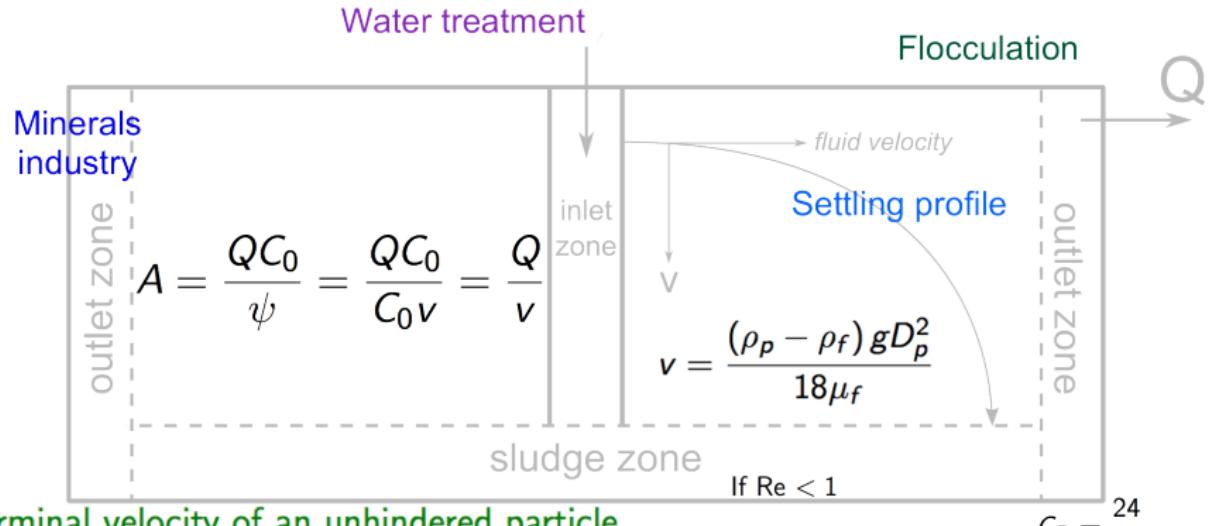
Download video: [Link](#) [801 M]

### 2014-4M3 Class 12A (19 Nov)

# Mechanical separations

<b>Part</b>	<b>Topic</b>	<b>Week number</b>
1	Sedimentation	2B, 2C, 3A, 3B
2	Particle size	3C, 3D
3	Centrifuges	3E, 4A, 4B
4	Filtration	4C, 5A, 5B
5	Cyclones	5C, 6A, 6B

## Sedimentation: 2B, 2C, 3A, 3B



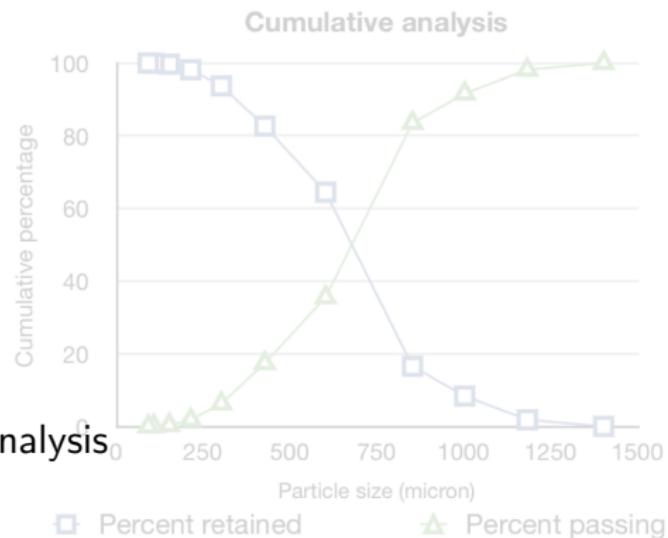
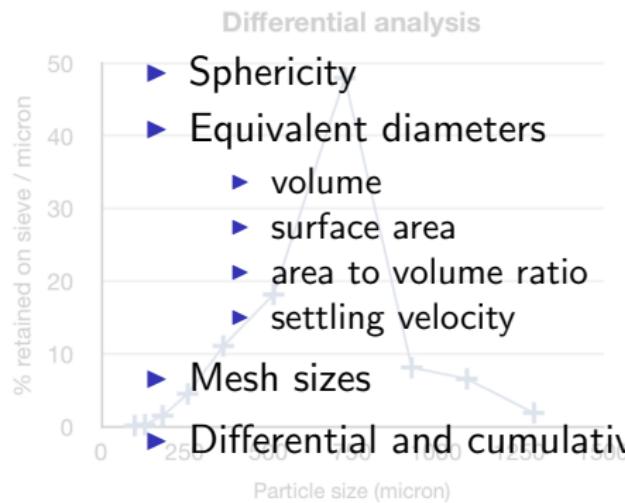
Terminal velocity of an unhindered particle

$$v = \sqrt{\frac{4 (\rho_p - \rho_f) g D_p}{3 C_D \rho_f}}$$

If  $1 < Re < 1000$

$$C_D = \frac{24}{Re} (1 + 0.15 Re^{0.687})$$

## Screens: 3C, 3D



## Centrifuges: 3E, 4A, 4B



liquid  
discharge

particle  
trajectory

- ▶ Many applications
- ▶ Tubular bowl and disk bowl

feed flow

$$Q_{\text{cut}} = \left( \frac{(\rho_p - \rho_f) g D_{p,\text{cut}}^2}{18 \mu_f} \right) \cdot (\Sigma) = v_{\text{TSV}} \cdot \Sigma$$

$$t_T = \frac{18 \mu_f}{D_p^2 (\rho_p - \rho_f) \omega^2} \ln \frac{r_2}{r_1}$$

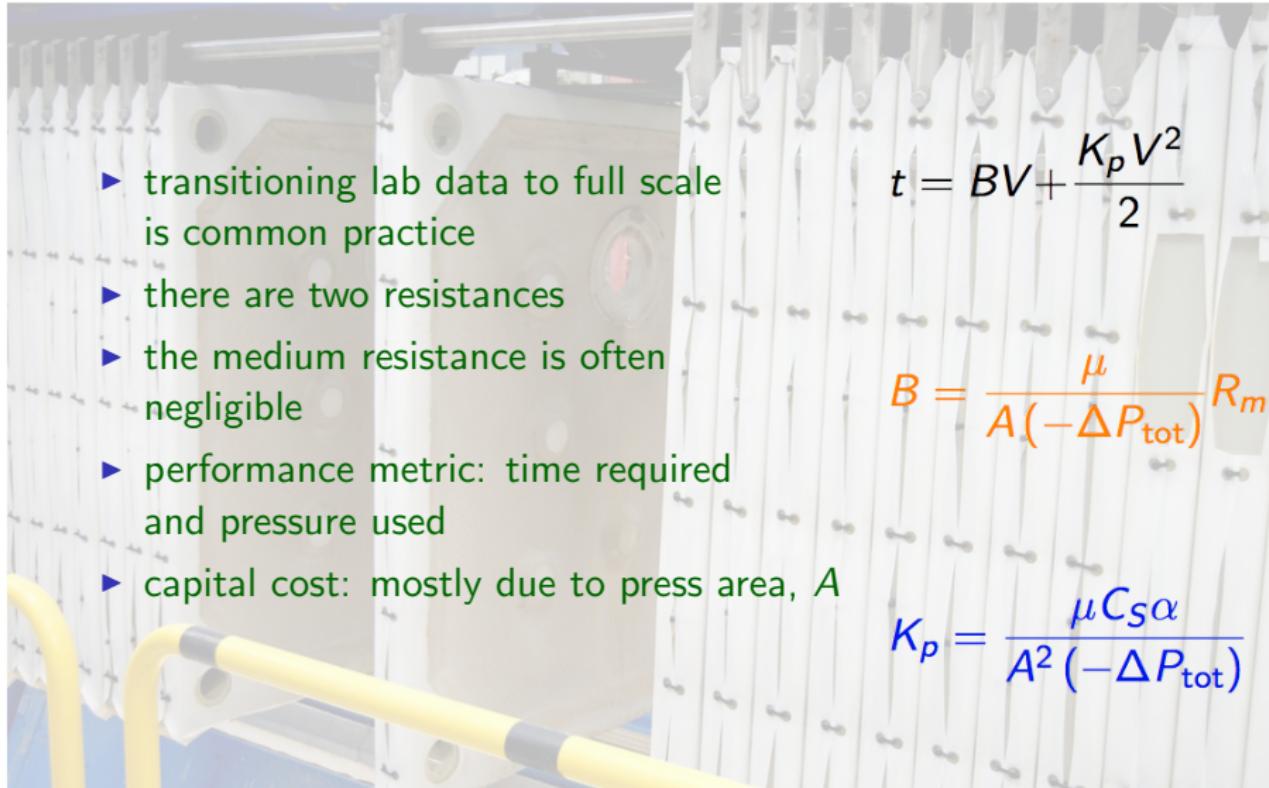
liquid  
surface

bowl  
wall

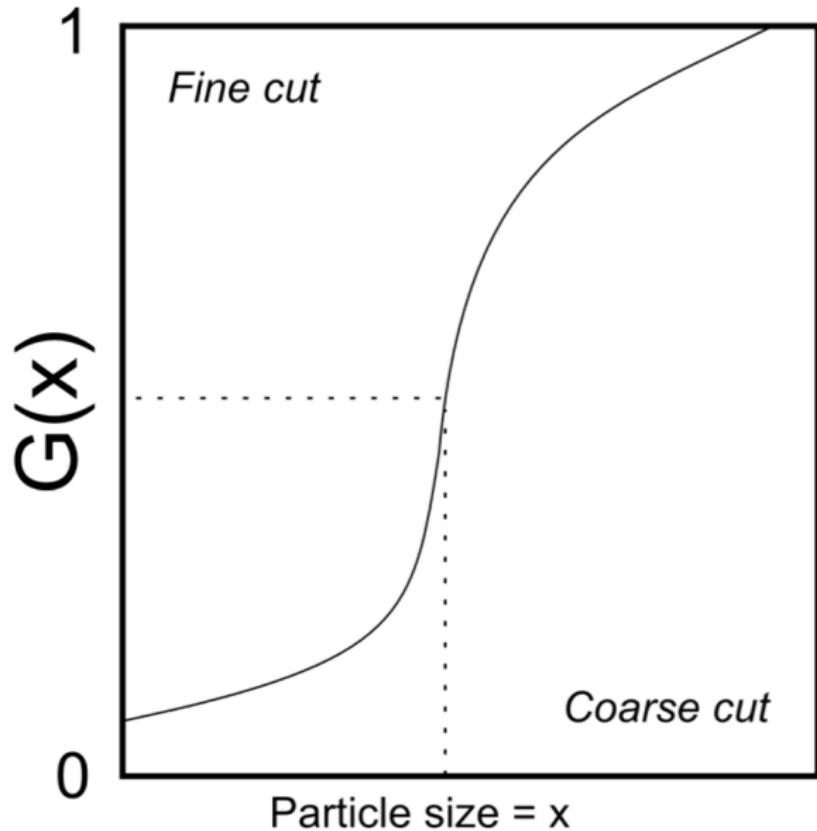
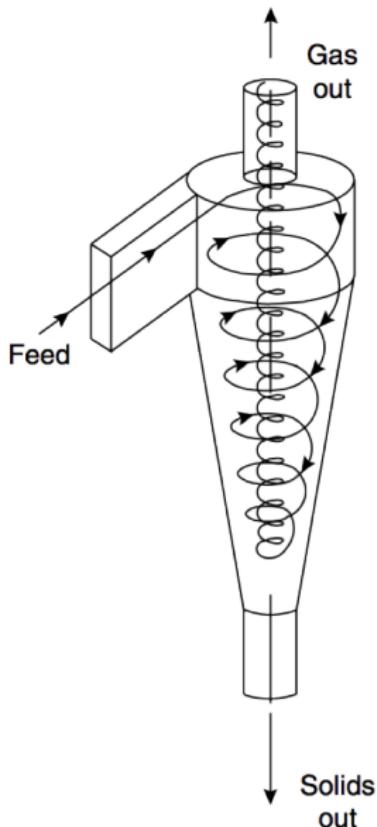
$r_2$

$$\frac{Q_{\text{cut},A}}{Q_{\text{cut},B}} = \frac{\Sigma_A}{\Sigma_B}$$

## Filtration: 4C, 5A, 5B



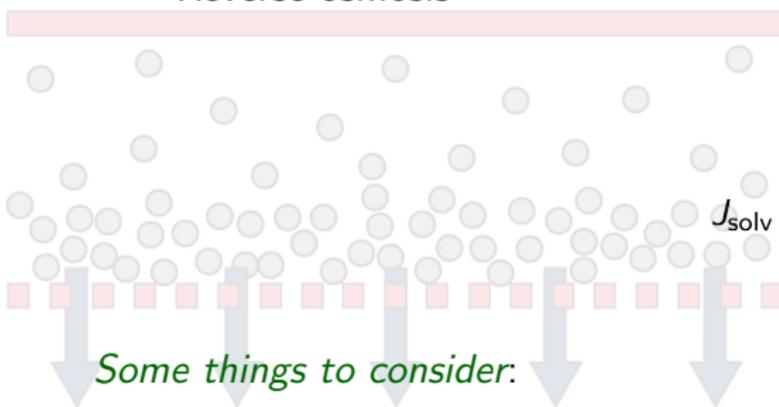
## Cyclones: 5C, 6A, 6B



# Membranes: 06C to 09A

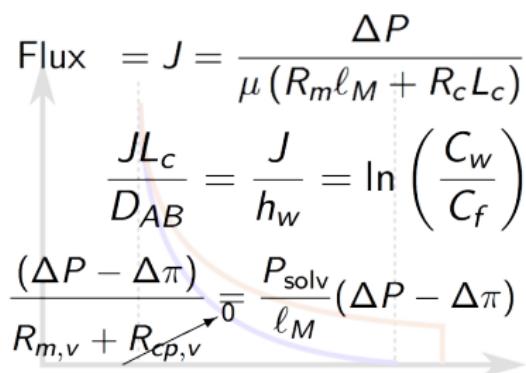
## We studied:

- ▶ Microfiltration
- ▶ Ultrafiltration
- ▶ Reverse osmosis



### *Some things to consider:*

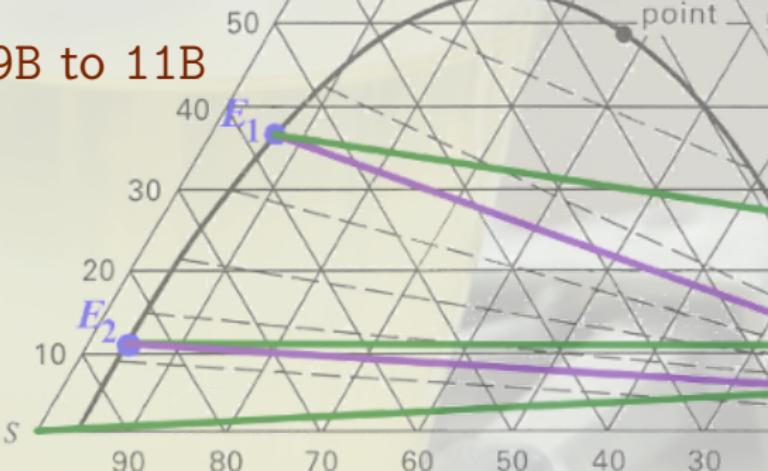
- ▶ What are typical LMHs,  $\Delta P$  and particle sizes retained?
- ▶ When can we set  $C_p \approx 0$ ?
- ▶ When can we disregard membrane resistance?
- ▶ How are permeances calculated?



# Liquid-liquid extraction 09B to 11B

## Plenty of new concepts

- ▶ ternary diagrams
- ▶ lever rule
- ▶ tie lines
- ▶ equilibrium
- ▶ solute, solvent, carrier
- ▶ mixer-settler, and other equipment alternatives
- ▶ extract, raffinate, distribution coefficient  $D_A = \frac{y_{E,A}}{x_{R,A}}$
- ▶ recovery and concentration
- ▶ units in sequence
- ▶ cross-current vs countercurrent units
- ▶ operating point  $P$

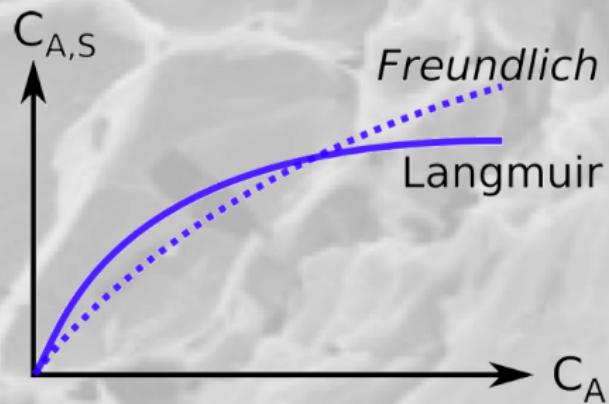


## Adsorption 11C to 12B

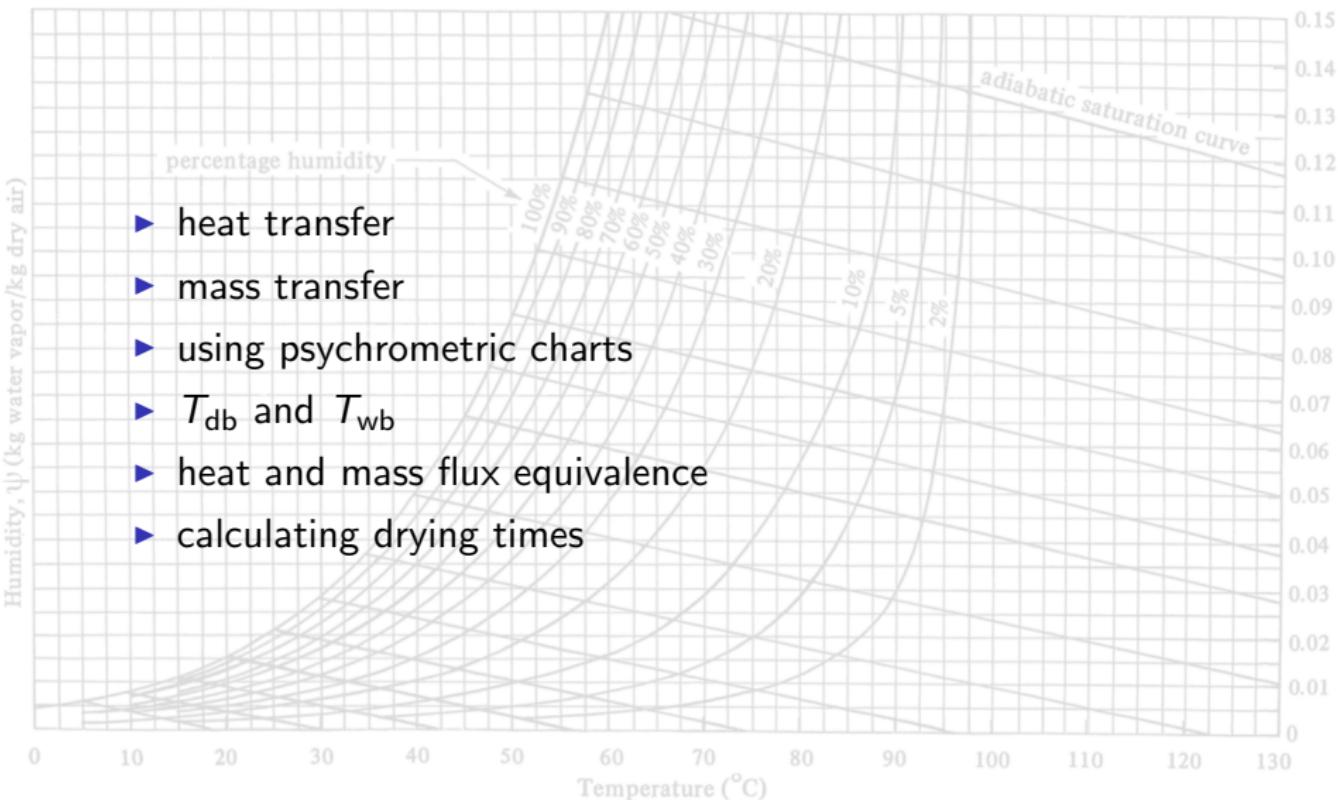
$$C_{A,S} = K (C_A)^{1/m}$$

- ▶ Langmuir and Freundlich isotherms
- ▶ Breakthrough
- ▶ MTZ
- ▶  $L_{LUB}$
- ▶ Bed mass balance

$$C_{A,S} = \frac{K_3 C_A}{1 + K_4 C_A}$$



# Drying 13A, 13B, 13C, 14A



## Common themes in all sections

- ▶ Separation factor =  $S_{ij} = \frac{x_{i,1}/x_{j,1}}{x_{i,2}/x_{j,2}}$
- ▶ Concentration of recovered compound in stream *i*
- ▶ Recovery =  $\frac{\text{mass of desired compound recovered in stream } i}{\text{mass of desired compound in the feed}}$
- ▶ Separating agents: mass (MSA) and energy (ESA)
- ▶ Type of capital costs (equipment) involved

## Take the following into account

For each separator we looked at, **please aim to:**

- ▶ understand the **physical principle** used in the separation
- ▶ know which **phases** are present and being separated?
- ▶ determine what affects the **unit's cost?**
- ▶ **identify variables** that can be used to fix a problem with the unit
- ▶ optimize an existing unit: increase throughput, boost recovery, aka “intensification”
- ▶ **repurpose** an existing unit for a similar, but different use.

## Other tips

- ▶ Understand the concepts being learned. My courses are not about applying the correct equation and solving.
- ▶ Read the questions carefully: they are worded precisely. Answer all parts of the questions.
- ▶ None of the final exam questions are going to be from the assignments (even with different values).
- ▶ Check that your answers are reasonable (can you really have a flow rate of  $1050 \text{ m}^3.\text{s}^{-1}$  through a pipe?)
- ▶ Computer questions in assignments: make sure you can repeat them by hand, where reasonable.

## Most important advice

- ▶ Treat the exam as a closed-book test: have a formula sheet for the equations, and understand all the concepts without referring to a textbook or notes
- ▶ Textbooks and other papers should be used to refer to as a backup only.

# Thank you

- ▶ It's been a long semester, **really busy**
- ▶ You've been the third round with 4M3 overhaul.
- ▶ Next year the class will be refocused a little more to water and energy topics
- ▶ But you have helped me tremendously with feedback about the notes and good questions in class and by email.
- ▶ Further comments? <https://evals.mcmaster.ca> or ...
- ▶ anonymously at  
<http://learnche.mcmaster.ca/feedback-questions>

**Thank you.**